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Flexible sensing probe for the simultaneous monitoring of neurotransmitters imbalance

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Abstract

Simultaneous detection of multiple neurotransmitters and their related activities is crucial for enhancing our under‑ standing of complex neurological mechanisms and disorders. In this study, we developed a fexible, high-sensitivity multi-electrodes array probe capable of concurrent detection of four neurotransmitters: dopamine, serotonin, acetylcholine and glutamate. The probe was fabricated on a polyimide substrate with 16 circular gold-flm electrodes. These electrodes were modified with PEDOT/GluOx and PEDOT/ChOx for enzymatic detection of glutamate and acetylcholine, and with rGO/PEDOT/Nafon for the detection of dopamine and serotonin. Our electrochemical sensor achieved sensitivities of 184.21 and 219.29 μ A/ μ M cm² for glutamate and acetylcholine, respectively, with limits of detection (LOD) of 0.0242 and 0.0351 μM within a concentration range of 0.1–100 μM. For dopamine and serotonin, the sensor showed sensitivities of 195.9 and 181.2 μ A/ μ M cm², respectively, with LOD of 0.4743 and 0.3568 μ M. This research advances the feld of neurochemical sensing and provides valuable insights into the balance of neurotransmitters associated with neurological disorders. These insights improve diagnostic and therapeutic strategies.

Keywords Flexible probe, Multi-neurotransmitters, Balance monitoring, Reduced graphene oxide, PEDOT:PSS

Introduction

The complexity of the human brain and its functions depends on the delicate balance and complex interplay of various neurotransmitters. Disruptions in neurotransmitter levels may lead to numerous neurological disorders, such as Alzheimer's disease, Parkinson's disease, anxiety, and bipolar disorder. Glutamate and choline, crucial excitatory neurotransmitters, play a pivotal role in maintaining the excitatory/inhibitory (E:I) balance in the brain $[1-3]$ $[1-3]$. Many studies have indicated that imbalances between neurotransmitters can lead to

various neurological disorders. Specifcally, an imbalance between cholinergic and dopaminergic activity in the striatum is implicated in conditions such as Parkinson's disease $[4-6]$ $[4-6]$ $[4-6]$. Therefore, there is a need for methods to monitor and quantify neurotransmitter levels to detect and address these imbalances.

The normal levels of DA in the brain are reported to range between 0.2 and 1 μ M, while 5-HT levels are documented at 0.28 to 1.14 μ M [[7,](#page-9-3) [8\]](#page-9-4). Glutamate and choline concentrations are described in the literature as ranging from 0.2 μM to approximately 20 and several µM, respectively [\[9](#page-9-5), [10](#page-9-6)].

Current methods for neurotransmitter detection, such as high-performance liquid chromatography and mass spectrometry, have been offered high accuracy but there are some limitations, such as long detection times, high costs, complex sampling preparation, and impossible for continuous monitoring $[11, 12]$ $[11, 12]$ $[11, 12]$ $[11, 12]$. These challenges necessitate the development of more efficient, cost-effective, and user-friendly techniques. Electrochemical sensing methods like diferential pulse voltammetry (DPV) and

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chronoamperometry have emerged as promising alternatives due to their simplicity, rapid detection capabilities, and high accuracy for precise neurotransmitter detection.

Recent advancements in materials have introduced carbon nanomaterials, particularly graphene, as exceptional tools for enhancing the sensitivity and specifcity of electrochemical biosensors [[13,](#page-9-9) [14](#page-9-10)]. While chemical vapor deposition is the primary method for fabricating graphene-based devices, this approach presents signifcant challenges due to the complexity and difficulty of device integration, as well as graphene's poor dispersion in solvents. Carbon nanotubes (CNTs) have also garnered signifcant attention from electrochemists for the development of improved sensing platforms, largely because of their electrocatalytic properties, which are particularly attributed to defects on the edge planes of the CNT structure [[15,](#page-9-11) [16](#page-9-12)]. Despite the availability of such materials, there remains a need for suitable materials that can selectively and efficiently detect target substances.

While signifcant progress has been made in developing sensitive measurement techniques for individual neurotransmitters, there remains a notable gap in the simultaneous monitoring of multiple neurotransmitters and their interactions [[17](#page-9-13)]. Addressing these issues, recent research has shifted towards developing multifunctional sensors capable of concurrent detection of multiple neurotransmitters. These advancements aim to provide a more comprehensive understanding of neurological functions and disorders.

In this study, the fexible sensing probe with a multielectrode array performed simultaneous multi-neurotransmitters detection. Modifed electrodes with reduced graphene oxide/PEDOT:PSS/Nafon (rGO/PP/NF) have shown signifcant potential in detecting DA and 5-HT with high sensitivity and selectivity. Since the graphene oxide can be deposited onto desired seed metal electrode, the electrochemically deposited graphene oxide could convert reduced graphene oxide by simple electrochemical reduction [[18\]](#page-9-14). In addition, the negatively charged nature of graphene oxide helps repel negatively charged interferents like uric acid (UA) and ascorbic acid (AA), thereby increasing the selectivity of the electrode [[19\]](#page-9-15). In the case of detection for choline and glutamate, the PEDOT:PSS was electroplated and the enzymes were simply immobilized on the electrodes.

In addition, monitoring the responses of 4 neurotransmitters (NTs) under induced imbalance conditions was conducted. This study not only allows for the simultaneous detection of various neurotransmitters but also ofers enhanced sensitivity, selectivity, and NTs' imbalanced level monitoring capabilities. This innovative approach has expectations for advancing our understanding of neurological disorders and improving diagnostic and therapeutic strategies.

Materials and methods

Materials

For fabricating the fexible sensor confguration, Polyimide (PI, VTEC 1388) was acquired from Richard Blaine International, Inc., Philadelphia, PA, USA. DNR-L300-30 was obtained from Dongjin, Seoul, Korea. AZ 9260 was acquired from AZ Electronic Materials, NJ, USA. Phosphate bufer saline (0.1 M PBS, pH 7.4) was obtained from Duksan General Science in Korea. DA, 5-HT, glutamate, choline, agarose, choline oxidase, o-phenylenediamine, glutaraldehyde, Nafon, ascorbic acid (AA), uric acid (UA), glucose, epinephrine (EP), norepinephrine (NE) and choline oxidase (100 units) were purchased from Sigma-Aldrich for electrochemical analysis. Glutamate oxidase (25 units) was from Yamasa, Japan.

Methods

The electrochemical performances of the sensors were evaluated by an Autolab (PGSTAT 302 N, NOVA software, Ecochemie, Utrecht, The Netherlands). Three electrode confgurations were used for CV, EIS with an Ag/AgCl reference, Pt counter, and modifed electrode (rGO/PP/NF, PP/ChOx, PP/GluOx) as a working electrode. The detection of simultaneous NTs wasmeasured using a µStat 8000 Multi Potentiostat/Galvanostat from DropSens company. For detection of neurotransmitters, bare Au electrodes near the working electrode were used as counter and reference electrodes.

The CV with potential limits of -0.2 and 0.8 V was performed with a scan rate of 100 mVs⁻¹, and the frequency range of EIS was from 1 to 105 Hz. The parameters of the DPV measurements were set as follows, i.e., the scan rate was 25 mV/s, the pulse width was 0.06 s, and the amplitude was 30 mV. The parameters of the amperometry measurements were set as 0.3 V. Fresh solution were prepared daily, and the sensors were kept in the dark at 4 °C to avoid the oxidation of DA. All of the experiments were conducted at ambient temperature.

The surface morphologies and elemental analyses of the electrode were evaluated respectively by scanning electron microscopy (SEM, Regulus8230).

Modifcation of rGO/PP/NF, PP/ChOx, and PP/GluOx surface on working electrodes

The schematic diagram of the electrodeposition procedure is displayed in Fig. [1.](#page-2-0) For preparation of rGO/PP/ NF electrode, GO was electrochemically deposited (1 μA current 540 s). After deposition, the composite was dried for 3 h. The obtained GO film on the surface was then reduced to rGO. To perform these reduction processes, a

Fig. 1 Schematic drawing of preparation of the fexible sensing probe with rGO/PP/NF, PP/ChOx, and PP/GluOx electrode

cyclic voltammetry technique was applied with a potential window between 0 and − 1.5 V at a scan rate of 50 mVs[−]¹ for 10 cycles in pH 4.01 bufer solution. After electrochemical reduction, the color of the rGO electrode was changed darker than GO electrode [[20\]](#page-9-16). Subsequently, the electrode was electrochemically coated in PEDOT:PSS (0.01 M EDOT and 0.1 M PSS in deionized water) solution for enhanced conductivity.

Finally, the rGO/PP was coated with 0.5 wt% Nafon (1 μ l) to repel unnecessary anions like AA and UA [[21\]](#page-9-17).

After that, for preparation of PP/ChOx and PP/GluOx electrodes, the working electrodes were electrodeposited by PEDOT:PSS. O-phenylenediamine was coated by cyclic voltammetry (0.2–0.8 V/50 mVs[−]¹ /20 cycle) for protection of enzyme activity, enhancing stability and improvement of sensor $[22]$ $[22]$. The electrode was covered with 0.125 wt% Glutaraldehyde for the crosslinking between surface and enzymes [\[23](#page-9-19)]. Glutamate oxidase (2 μl) was drop-casted onto the electrode, enabling specifc measurement of glutamate, while choline oxidase

drop-casting facilitated the measurement of choline exclusively. The modified electrodes were stored in a dark room for 24 h and the Nafon was drop casted for the protection layer of enzyme before use.

Results and discussion

Fabricated fexible sensing probe

Fig. S1. shows a simple microfabrication process of the fexible sensing probe. Initially, a polyimide (PI) layer (20 μm) was spin-coated onto a 4-inch dummy wafer. Following curing, a negative photoresist was spin-coated atop the PI layer. Subsequently, Cr/Au (10/100 nm) layers were deposited using an e-beam evaporator. After the lift-off process, a second layer of photosensitive PI was spin-coated and cured to serve as the insulation layer. A laser dicing machine was employed to trim the perimeter of the sensor. Finally, the fexible sensor was easily detached from the wafer.

Figure [2](#page-3-0) shows the conceptual drawing, photograph image of the fabricated fexible sensing probe, and microscopy image for the rGO/PP/NF, PP/ChOx and $PP/GluOx$. The width of the shaft is 2.6 mm, and the length is 21.5 21.5 mm (Fig. $2a$). The electrode array is composed of 16 circular electrode sites designed for multineurotransmitter (NT) detection, with an electrode radius of 300 μ m. For detection of neurotransmitters, bare Au electrodes near the working electrode were used as counter and reference electrodes. This image depicts the optical microscope (OM) view after sequentially depositing rGO/PP/NF onto the electrodes on the probe shank (Fig. [2b](#page-3-0)). The electrode color becomes progressively darker with each deposition step from GO to rGO and then to rGO/PEDOT. The lower image shows the OM view after preparing PP/ChOx and PP/GluOx, confrming the successful attachment of the two types of enzyme particles (Fig. [2](#page-3-0)c).

Morphological analysis

Figure [3](#page-4-0) presents SEM images of GO, rGO, rGO/PP, and $rGO/PP/NF$. The deposited layers of GO in Fig. [3](#page-4-0)a exhibit a single or few-layered microstructure with surface wrinkles, likely due to the presence of numerous functional groups such as hydroxyl and carboxyl groups at the edges, and carboxyl and epoxide groups within the inner structure. In Fig. [3b](#page-4-0), the rGO electrodes display a morphology with abundant wrinkles and fuctuations. These fluctuations are crucial for maintaining the thermodynamic stability of graphene due to its 2D crystal structure. Notably, the surfaces of rGO/PP show diferent structures compared to rGO, appearing to be covered by uniformly distributed PEDOT:PSS (Fig. [3](#page-4-0)c). Furthermore, the surface of rGO-PP/NF appears similar with rGO/PP, with the same wrinkled structure (Fig. $3d$). This suggests that the Nafon layer did not exert any signifcant infu-ence on the existing layer [[24\]](#page-9-20). In Fig. [3](#page-4-0)e, the pristine PEDOT:PSS layer that was polymerized onto the thin Au electrode exhibited a homogeneous distribution of the nanoparticles like grain of sand. After immobilization of enzymes, there are enzyme particles on the PEDOT:PSS surface. (Fig. [3e](#page-4-0) and f).

Electrochemical properties

To assess the electrochemical performance of the modifed electrodes, we conducted measurements of interfacial impedance for Au, GO, rGO, and rGO/PEDOT:PSS electrodes in a 0.1 M PBS solution (pH 7.4), as depicted in Fig. [4a](#page-4-1). At a frequency of 100 Hz, the recorded interfacial impedance values for Au, GO, rGO, and

Fig. 2 a Microscopic image of the fexible sensing probe; **b** Photograph of the fabricated fexible sensing probe with rGO/PP/NF; **c** fabricated fexible sensing probe with PP/ChOx and PP/GluOx

Fig. 3 SEM images of the fabricated **a** GO, **b** rGO, **c** rGO/PP, **d** rGO/PP/Nafon, **e** PEDOT:PSS, **f** PP/ChOx, and **g** PP/GluOx

Fig. 4 Comparison of the (**a**) interfacial impedance and (**b**) cyclic voltammograms of Au, GO, rGO, and rGO/PEDOT:PSS composite; comparison of the (**c**) interfacial impedance and (**d**) cyclic voltammograms of the Au, PEDOT:PSS, PEDOT:PSS/ChOx, and PEDOT:PSS/GluOx

rGO/PP electrodes were 8826.68, 6506.12, 2066.53, and 213.613 $Ω$, respectively. Remarkably, the rGO/ PP electrode exhibited significantly enhanced interfacial impedance compared to the other electrodes, owing to its superior conductivity and low resistance to charge transfer [[24](#page-9-20)]. Additionally, after deposition of PEDOT:PSS, It showed amplifed current response, indicating an augmentation of electrochemically active sites [[25](#page-9-21)].

In the measured CV curve, the CSC value, representing the accumulated charges, increased sequentially in the order of $Au < GO < rGO < rGO/PP$, with corresponding values of 0.07, 0.925, 1.694, and 8.21 mC/cm², respectively (Fig. [4](#page-4-1)b).

Cyclic voltammetry measurements confrmed that the electroactive area of the electrode increased with each deposition [[26\]](#page-9-22). After depositing PEDOT:PSS, the impedance decreased compared to the bare gold electrode (Fig. [4c](#page-4-1)). However, the impedance increased after immobilizing each of the choline and glutamate enzymes.

DA

35

34

a)

Simultaneous detection of multi‑NTs

Figure [5a](#page-5-0) shows the DPV responses to simultaneous injections of DA and 5-HT across a concentration range of 1 to 50 μ M. Two distinct peaks are observed, corresponding to DA at approximately 0.2 V and 5-HT at 0.48 V. The distinct oxidation potentials of DA and 5-HT lead to their separate peaks in the DPV spectrum. The fexible sensing probe with rGO/PP/NF exhibited linear current response with increasing concentration of DA and 5-HT. The correlation coefficients for DA and 5-HT with respect to their concentrations are 0.91 and 0.98, respectively. The sensitivities for DA and 5-HT are 195.9 and 181.2 $\mu A \mu M^{-1}$ cm², respectively, with limits of detection (LOD) of 0.4743 and 0.3568 μM.

DA

35

34

 R^2

 $= 0.91$

 $Slope = 0.67$

 $b)$

of DA and 5-HT; **c** Choline/Glutamate (Amperometry, 0.1–100 μM) at the fabricated fexible sensing probe with PP/ChOx, and PP/GluOx electrodes; **d** Calibration curve plot of choline and glutamate

The amperometric current responses and calibration curve to the various choline and glutamate $(0.1 \text{ to } 100 \mu M)$ concentrations are shown in Fig. [5b](#page-5-0). The correlation coefficients for choline and glutamate are 0.99 and 0.97, indicating a strong correlation between concentration and current response. The sensitivities for choline and glutamate are 184.21 and 219.29 μA μ M⁻¹ cm², respectively, with LODs of 0.0242 and 0.0351 μM.

Supplementary Fig. S2 indicates that the current responses for the four neurotransmitters are relatively well matched with individually measured ones. This result means that four diferent neurotransmitters can detect independently without crosstalk among them.

In order to check the response in circumstance with NT imbalance, we checked NT responses under higher concentrated conditions of the arbitrary NT (Fig. [6](#page-6-0)). All the injected NT levels were set-up based on clinically relevant concentrations found in cerebrospinal fuid [\[28\]](#page-9-24).

In Fig. [6a](#page-6-0), a high concentration of DA $(10 \mu M)$ was injected, while other neurotransmitters were maintained at normal levels. Despite the increment of DA concentration, the responses of the other neurotransmitters remained stable, indicating their responses are independently measured. In addition, the Fig. [6](#page-6-0)b–d showed the responses to the similar circumstance. In all cases, the high concentration of arbitrary single neurotransmitter did not afect to the signifcant changes in the response of the other neurotransmitters. This finding demonstrates that the sensor can stably and reliably detect individual neurotransmitters even in a mixed environment.

Selectivity and stability

The presence of four types of neurotransmitters and other interfering species can lead to mixed response currents due to their closely spaced oxidation potentials. To investigate the selectivity of the rGO/PP/NF electrode, DPV was employed to measure the oxidation current response of DA at 100 μM. Glutamate (100 μM), choline (100 μM), ascorbic acid (100 μ M), and uric acid (100 μ M) in 0.1 M PBS (pH [7](#page-8-1).4) were tested in Fig. 7a. The response currents in the presence of interfering substances were negligible compared to that of 100 μ M DA, demonstrating the high antiinterference capability of the rGO/PP/NF to the glutamate, choline, ascorbic acid, and uric acid.

The reproducibility of the flexible sensing probe with rGO/PP/NF was investigated by using five electrodes. As depicted in Fig. [7](#page-8-1)b, the DPV-measured peak currents for DA oxidation in 0.1 M PBS (pH 7.4) were similar, ranging within \pm 1.00 μA of the average value of 32.584 μA.

Figure [7c](#page-8-1) illustrates the selectivity test using the fexible sensing probe with PP/GluOx. When 100 μM of serotonin, dopamine, ascorbic acid, and uric acid were injected, no signifcant responses were observed. However, the sensor clearly detected 100 μM and 1000 μM of glutamate. The reproducibility of the flexible sensing probe with PP/ ChOx and PP/GluOx was evaluated as shown in Fig. [7](#page-8-1)d. The current responses for 10 μ M of choline and glutamate were consistent and maintained similar levels. The choline sensors showed an average current of 0.030948 μA with a variability of \pm 6.71%, while the glutamate sensors exhibited an average current of 0.0283 μ A with a variability of \pm 6.4%. These consistent current responses across the five sensors indicate high reproducibility.

Conclusion

In this study, we successfully introduced and validated a high-sensitivity flexible sensing probe capable of simultaneously DA, 5-HT, choline, and glutamate. The sensor array, modifed with rGO/PP/NF and PEDOT as working electrodes, demonstrated excellent individual NT sensing properties and reliable simultaneous detection in scenarios involving NT imbalances.

The probe's sensitivities for DA and 5-HT were 195.9 and 181.2 $\mu A \mu M^{-1}$ cm², respectively, within a concentration range of $1-50$ μM. The LOD for DA and 5-HT were 0.4743 and 0.3568 μM, respectively. For choline and glutamate, within a range of $0.1-100 \mu M$, the probe achieved sensitivities of 184.21 and 219.29 $\mu A \mu M^{-1}$ cm², with LODs of 0.0242 and 0.0351 μM, respectively.

The selectivity tests verified that the presence of 100 μ M interferents did not signifcantly afect the sensor's performance, thereby ensuring reliable NT measurement. Reproducibility tests validated the probe's feasibility for use with fve sensors, maintaining consistent performance over time.

0.1 M PBS is conventionally and widely used as a bufer solution in laboratory experiments due to its ability to maintain consistent pH and ionic strength, which are critical for ensuring the stability of both the sensors and the target analytes. Unlike the PBS buffer, the applicability in complex biological samples, such as extracellular fuid or brain tissue, should be further investigated about its durability, stability, and feasibility. Nevertheless, we are expecting that the developed sensing probe will be operated well even in the biological samples due to Nafon coating as a negative ion repelling and protection layer and the DPV detection technique for the selective and independent multi-NTs determination.

(See fgure on next page.)

Fig. 6 Crosstalk monitoring at the simultaneous determination of the DA/5-HT and choline/glutamate to the induced imbalance situations; **a** imbalance of DA, **b** imbalance of 5-HT, **c** imbalance of choline, **d** imbalance of glutamate

Fig. 6 (See legend on previous page.)

Fig. 7 Selectivity and stability of the fabricated fexible sensing probe for determination of DA/5-HT and choline/glutamate; **a** comparison of DPV responses of the DA to the adding of diferent interfering species (100 μM DA and 100 μM DA with interferents (100 μM Choline, Glutamate, UA, AA); **b** reproducibility of 5 diferent fexible sensing probe with rGO/PP/NF; **c** amperometric response of the fexible sensing probe with PP/GluOx electrode to the diferent interfering species (100 μM 5-HT, DA, AA, UA) and 100/1000 μM glutamate; **d** reproducibility of 5 diferent fexible sensing probe with PP/ChOx and PP/GluOx

This work has significant potential for applications in neuroscience research and clinical diagnostics, ofering a tool for monitoring neurotransmitter dynamics with high sensitivity.

Supplementary Information

The online version contains supplementary material available at [https://doi.](https://doi.org/10.1186/s40486-024-00211-3) [org/10.1186/s40486-024-00211-3](https://doi.org/10.1186/s40486-024-00211-3).

Additional fle 1.

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Author contributions

Experiments were designed by Y.J.L. Fabrication process of the sensors was performed by Y.Z. Characterization of the fexible sensing probe were performed by H.B.C. H.B.C, H.Y.Y., and Y.J.L. wrote the main manuscript text and prepared all fgures. Y.J.L. supervised all aspects of this work. All authors reviewed the manuscript and have given approval to the fnal version of the manuscript.

Data availability

No datasets were generated or analysed during the current study.

Declarations

Competing interests

The authors declare no competing interests.

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