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A review of silicon microfabricated ion traps for quantum information processing

Dong-II "Dan" Cho^{1*}, Seokjun Hong¹, Minjae Lee¹ and Taehyun Kim²

Abstract

Quantum information processing (QIP) has become a hot research topic as evidenced by S. Haroche and D. J. Wineland receiving the Nobel Prize in Physics in 2012. Various MEMS-based microfabrication methods will be a key enabling technology in implementing novel and scalable ion traps for QIP. This paper provides a brief introduction of ion trap devices, and reviews ion traps made using conventional precision machining as well as MEMS-based microfabrication. Then, microfabrication methods for ion traps are explained in detail. Finally, current research issues in microfabricated ion traps are presented. The QIP renders significant new challenges for MEMS, as various QIP technologies are being developed for secure encrypted communication and complex computing applications.

Keywords: Microelectromechanial System (MEMS); Microfabrication; Ion traps; Quantum information processing (QIP); Quantum computing

Introduction

Quantum information processing (QIP) is a novel information processing method based on quantum mechanics [1-3], and uses two quantum states in a quantum system as a basic unit of information, instead of two voltage levels in conventional information processing based on electronics. This basic unit is called "qubit", an abbreviation for quantum bit. The information stored in a single qubit exists in a superposition of two quantum states which indicates an arbitrary linear combination of two orthonormal basis. Since a single qubit can occupy either of two states simultaneously, N qubits can represent 2^N states of information. Moreover, using a quantum teleportation process [4], two gubits can provide the same measurement results, regardless of the distance between the qubits. Based on these phenomena in the quantum regime, QIP is expected to achieve noticeable increases in the speed in information processing problems. Therefore, many QIP applications such as quantum communication [5-7], quantum computer [8-12], and quantum simulator [13-15] have been proposed and are being actively researched.

For the physical implementation of the qubit, a quantum system which is sufficiently isolated from their surroundings and can be individually manipulated is required. Individual manipulation means gubits are initializable, controllable and measureable. A single atomic ion confined by a physical platform which is called "ion trap" satisfies the requirements [16-19]. Thus the ion trap has become one of the leading technologies among the various qubit platforms including superconducting circuit [20-22], optical lattice [23,24], nuclear magnetic resonance (NMR) [25,26], and quantum dot [27,28]. The ion trap was initially developed by Wolfgang Paul and Hans Georg Dehmelt who are the co-winners of the Nobel Prize in Physics in 1989. Since Cirac and Zoller have proposed using trapped ions as a physical implementation of qubit [16], the feasibility of ion qubits has been verified through many experiments [19,29,30]. Recently, in 2012, Serge Haroche and David Wineland received the Nobel Prize in Physics owing to the measurement and manipulation of individual quantum systems, using cavity quantum electrodynamics (QED) and ion traps, respectively. There has been several review articles on the subject of quantum information processing [18,31-34].

Although the earlier Paul traps were constructed by conventional precision machining method and careful manual assembling, with the advances in MEMS, recent

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ion traps are based on silicon microfabrication technologies. The basic principles of ion traps are presented in Types of ion trap section. Then, Development history of Paul trap section discusses a history of the Paul trap, which is the type of ion traps mainly covered in this paper. In MEMS-based microfabrication section, two MEMS microfabrication methods for ion traps are explained. Finally, the current issues and the future development directions of microfabricated ion traps are presented in Future directions section.

Types of ion trap

An ion trap is a device which can trap charged particles in space by using electric or electromagnetic fields. Trapping a charged particle with static potential alone is impossible because the static potential (φ) obeys one of Maxwell's equations $\nabla^2 \varphi = 0$ [35]. Wolfgang Paul used an oscillating electric field together with the static electric field [36], and Hans Georg Dehmelt added a magnetic field to the static electric field to trap a positive ion [37]. The ion traps built by Paul and Dehmelt are called "Paul trap" and "Penning trap" [38] respectively. In this paper, we cover only the Paul trap, because the Paul trap is currently widely used for QIP applications.

Figure 1(a) shows the structural schematic and trapping principles of the Paul trap, which is composed of a ring-shaped hyperbolic electrode and two endcap electrodes located at the top and bottom of the ring electrode. This type of ion trap is called the "ring trap" compared with the next generation traps which are called the "linear trap" and the "surface trap". The details of the linear and surface traps are discussed in Development history of Paul trap. Figure 1(b) shows the ring trap made by Wolfgang Paul [39]. In the ring trap, a charged particle is confined in both radial and axial directions by the RF voltage applied to the hyperbolic electrode, and the endcap electrodes function as an RF ground [40]. Typically, the magnitude of the RF voltage is up to hundreds of volts and the frequency of the RF voltage is from tens to hundreds of MHz.

Development history of Paul trap Ring trap

In the early stages of ion trap researches, the ring type Paul trap was used for experiments concerned in fundamental physics such as frequency standards [41] and mass spectroscopy [42,43]. Ring traps can be easily constructed because of its simple structure, but has a drawback in trapping large numbers of ions because a potential minimum exists at a specific point and difficult to be expanded to a 3-D space.

Linear trap

To store more ions, a new ion trap called the "linear trap" was built by Prestage *et al.* [44]. The first linear trap held approximately 20 times the number of ions as that of a ring trap. Figure 2(a) shows the electrode

structure and electric fields in the radial plane of a linear trap. The hyperbolic electrode of the ring trap is replaced by four rods. Sectors of the circular rods form a hyperbolic shape, and an RF voltage is applied to two opposite rods with the other rods grounded. The pseudopotential generated by the RF voltage confine the ions in the radial direction, and the radial position of the ions is at the center of the four rods. Because the RF null is expanded along the axial direction of the rods, the pseudopotential from the RF voltage cannot confine ions in the axial direction. To

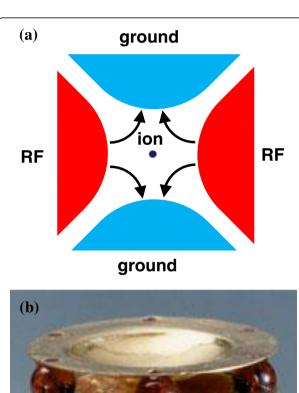




Figure 1 Schematic and picture of "ring trap". (a) Schematic of a ring trap. The red and blue regions indicate RF and DC electrodes respectively. The curved arrows denote the direction of electric field when RF voltage is positive. (b) A ring trap made by Wolfgang Paul [39]. Ion is trapped inside the ring structure.

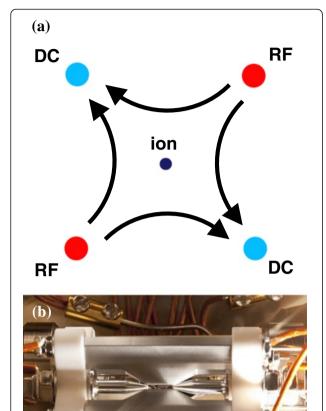


Figure 2 Schematic and picture of "linear trap". (a) Schematic of a 4-rod linear trap. The red and blue circles indicate RF and DC electrodes respectively. The curved arrows denote the direction of electric field when RF voltage is positive. **(b)** A blade type linear trap of Innsbruck group [48]. (© R. Blatt, University of Innsbruck).

confine trapped ions in the axial direction, two endcap electrodes are located at both ends of rod electrodes, and DC voltages are applied to the endcap electrodes to confine the ions axially [45,46].

Cirac and Zoller [16] proposed using trapped ions as a physical implementation of quantum computation. Since then, many research groups have been using linear traps in their QIP experiments. Most of the groups have developed their own linear traps using precision machining and assembling techniques. Each research group has a different electrode structure. Some typical electrode structures include rods [45,47], blades [48,49] and sheets [50]. Figure 2(b) shows a blade type linear trap of the Innsbruck group [48]. Many ion trap research groups are still using a variation of these 4-rod linear traps. In general, when compared to the surface traps (explained in the following sub-section) the 4-rod linear traps have

a higher trap depth, which in turn provide a longer ion life time and more stable trapping of ions. However, the linear traps do not offer the design freedom of the surface traps, and currently more research efforts are being expended to the surface traps.

Surface trap

To implement more complicated quantum operations, more ions that can be manipulated in a common motional mode (which refers to the collective oscillation of the whole ion string) should be trapped. Therefore the idea of integrating multiple ion trap arrays in a single ion trap chip was proposed [12,51]. The ion trap chip integrated with multiple ion trap arrays is divided into different regions, as an operation region in which the quantum

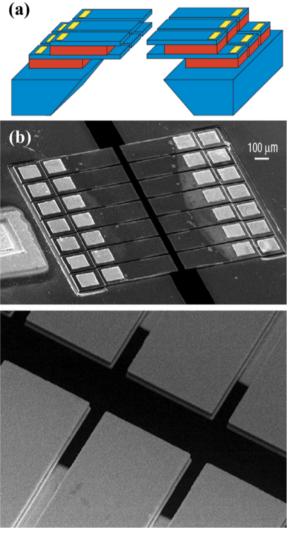


Figure 3 Schematic and picture of the first ion trap fabricated by a semiconductor fabrication process. (a) Schematic of the ion trap [53]. (b) Scanning electron micrograph of the ion trap [53].

operations are held, a memory region that stores ions conserving qubit states, and a region for loading ions.

Scalable microfabrication technologies were applied for the implementation of these large scale integrated ion traps, and the first microfabricated ion trap chip implemented the 4-rod ion trap configuration using a gallium arsenide (GaAs) based semiconductor fabrication process as shown in Figure 3 [52,53]. In this method, however, the upper and lower electrode layers are separated by an

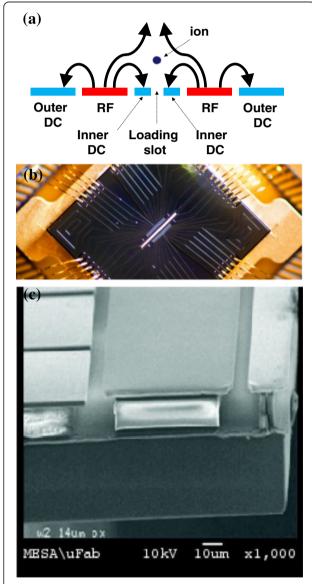


Figure 4 Schematic and pictures of "surface trap". (a) Schematic of a surface trap. The red and blue rectangles indicate RF and DC electrodes respectively. The curved arrows denote the direction of electric field when RF voltage is positive. Note that all electrodes are laid on the same plane. **(b)** Optical image of the surface trap of the Sandia National Laboratory (SNL) group, which has a double metal layer on Si substrate [58]. **(c)** Scanning electron micrograph of the surface trap of the SNL group [59].

epitaxially grown aluminium gallium arsenide (AlGaAs) layer, and the vertical distance between the electrodes is limited to a few micrometers whereas the horizontal distance is relatively large at 60 micrometers because of the laser access. This structural asymmetry results in a low radial confinement, which in turn results in a fast ion loss.

To overcome the limitation of implementing a 3-D structure using essentially 2-D fabrication techniques, a breakthrough in the 2-D planar ion trap where all electrodes are laid in the same plane was proposed [54-57]. This is more suitable for the silicon-based MEMS microfabrication technology. These 2-D ion traps are called the "surface traps". The surface traps have advantages in the scalability, and are now more widely used by many research groups. Figure 4(a) shows the structure and the principle of the surface trap. All electrodes are fabricated in the same plane. A radial confinement is controlled by two planar RF electrodes, and the position of an RF null where ions are trapped is placed above the two RF electrodes. In the 4-rod linear trap, the radial RF null is fixed at the center of the four rods. In the surface trap, the RF null position is above the substrate plane, and the height can be changed as a function of the width of the RF electrodes and the distance between RF electrodes. In most surface trap experiments, the laser path is parallel to the trap surface. For cooling trapped ions by a laser parallel to the trap surface, the principal axis in a radial plane should be rotated. The rotation of principal axis can be realized by varying the widths of the two RF electrodes, or by the addition of inner DC electrodes where an asymmetric set of DC voltages is applied to [56,57].

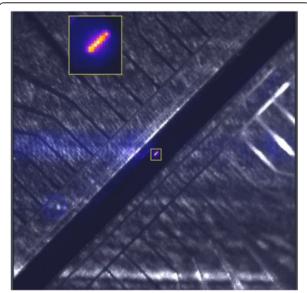


Figure 5 ¹⁷⁴Yb⁺ ion string trapped on a surface trap fabricated by our group [62]. The image is acquired by electron multiplying charge coupled device (EMCCD). The image of electrodes were taken separately and overlaid for clarity.

Multiple DC electrodes are also fabricated outside the RF electrodes. These outer DC electrode function as the RF ground. By applying different control voltages to these outer DC electrodes, trapped ions can be axially confined. Furthermore, by applying time-varying control voltages to the outer electrodes, the trapped ions can be moved along axis or shuttled.

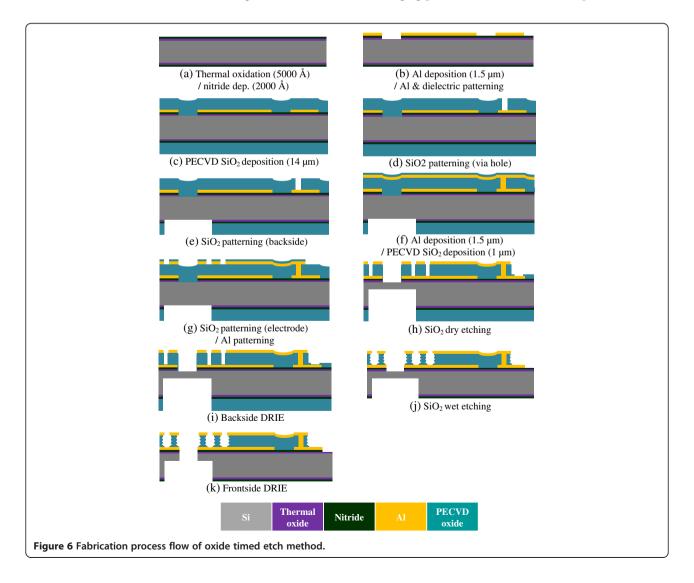
Figure 4(b) and 4(c) show a surface trap developed by the Sandia National Laboratory (SNL) group [58,59], which has a double metal layer on Si substrate [60-62]. The surface trap of the SNL group has been used by many research groups, including those at UC Berkley, Duke University, and Georgia Tech Research Institute through the "Ion Trap Foundry Program", to successfully demonstrate various quantum experiments [63-65]. We also developed a similar trap chip with optimized shapes, and successfully trapped a string of ¹⁷⁴Yb⁺ ions as shown in Figure 5 [62]. In Figure 5, a magnified view of the chip layout is also shown. The DC control electrodes are segmented and laid

vertically to the RF rails to generate the axial potentials with an appropriate shape for an axial confinement or a shuttling.

In addition to the Si-based surface traps mentioned in the above, surface traps with a single metal layer on a nonconductive substrate, fabricated by patterning Au electrodes on quartz or sapphire substrates [54,66-69] have been reported. A surface trap has also been fabricated on printed circuit boards [70-72].

MEMS-based microfabrication

Although many results have been reported on trapping ions with MEMS-fabricated traps, the process details to fabricate the trap chips are very scarce in the literature. Fabricating ion traps requires thick dielectric films to withstand several hundred volts of RF voltages. However, the dielectric layer should be as invisible as possible as seen from the RF null point where ions are trapped, since dielectric charging phenomena can alter the null position and can



induce the micromotion of trapped ions. In this section, we introduce two fabrication methods developed by us.

The first method is explained in the below. First, a silicon wafer is cleaned by using the Piranha solution and thermally oxidized to form a 5000-Å SiO₂ dielectric layer. A 2000-Å silicon nitride film is deposited by a low pressure chemical vapor deposition (LPCVD) process to protect the thermal SiO₂ layer during buffered oxide etching (BOE) at the end of process. These dielectric layers have total thickness of approximately 0.7 µm and must be sufficiently thick to prevent a breakdown between the bonding pads and the silicon substrate (Figure 6(a)). A 1.5-µm thick aluminum ground layer is deposited on the silicon nitride film by sputtering and dry etched to provide a ground plane for RF shielding and to form wire bonding pads of DC electrodes (Figure 6(b)). A 14-µm thick SiO₂ film is deposited by plasma enhanced chemical vapor deposition (PECVD) in several layers to control residual stress (Figure 6(c)). In this case, each SiO₂ layer is 3–4 μm thick and deposited alternately on both sides of the wafer to prevent the residual stress to build up. The PECVD-SiO₂ layer on the frontside of the wafer is dry etched to fabricate via holes (Figure 6(d)). After the dry etching of the frontside, the PECVD-SiO₂ layer on the backside of the wafer is dry etched to provide an oxide hard mask for deep reactive ion etching (DRIE) (Figure 6(e)). In the dry etching process of the 14-µm thick SiO2, the AZ 4620 (Clariant Corporation) photoresist (PR) is used as the etch mask. An additional 1.5-µm thick aluminum layer which is used as electrodes is deposited through a sputtering process (Figure 6(f)). The electrode layer also covers the sidewalls of the etched via holes where the bonding pads and electrodes are electrically connected at. The electrode layer and the PECVD-SiO₂ layer are patterned and define the electrodes and oxide pillars, respectively (Figure 6(g), (h)). The silicon substrate is etched 450 µm by a DRIE process from the backside of the substrate (Figure 6(i)). The overhang structures are fabricated by the oxide wet etching process using a buffered hydrogen fluoride (BHF) solution (Figure 6(j)). Because the PECVD oxide is deposited in several steps, the sidewall profile becomes jagged. The wet etching time must be precisely controlled. Some overhang is required to reduce the dielectric layer exposure to the trapped ions, but too large an overhang can result in a long cantilevered top Al layer, which can bend with applied high voltages. After the oxide wet etching process, the slit-shaped ion loading hole is fabricated by a DRIE process, and the fabrication is finished (Figure 6(k)). Figure 7 shows the fabrication results of the ion trap fabricated by this oxide timed etching method described in the above.

This timed etch method is simple but the electrode overhang dimensions are difficult to control. An alternative fabrication method is possible using a sacrificial

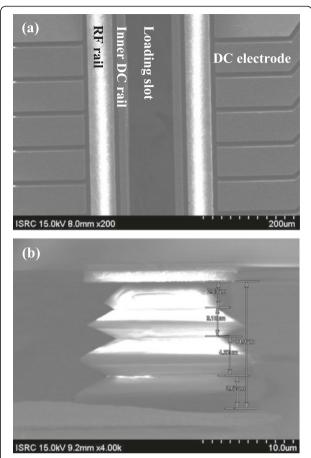
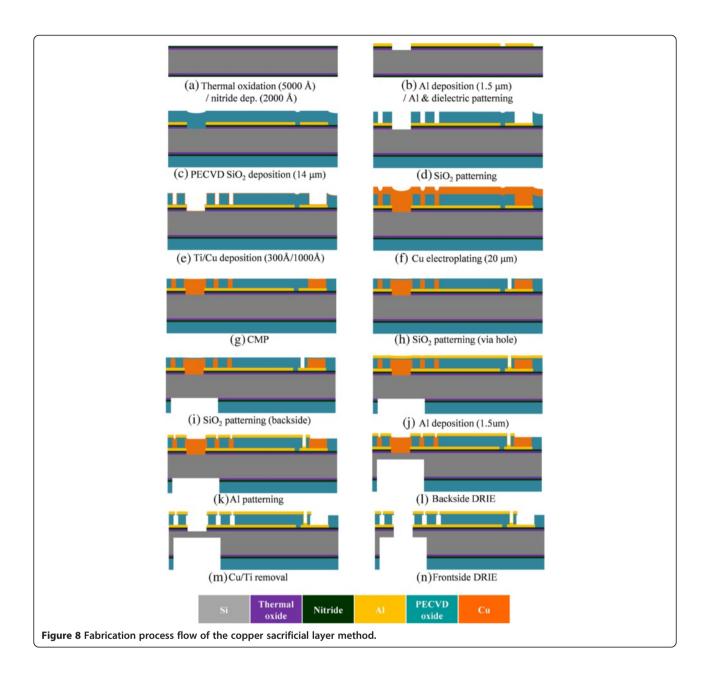


Figure 7 Micrographs of ion trap fabricated by oxide timed etch method. (a) Scanning electron micrograph which shows the top structure of the ion trap. (b) Scanning electron micrograph of the cross section, showing the jagged edges and electrode overhang.

material. The fabrication process is the same as the method described in the above up to the step of dry etching the 14-µm thick PECVD-SiO₂ (Figure 8(d)). In this study, copper is used as a sacrificial material. The copper technology is selected because it is inexpensive and readily available from the advances in the through silicon via (TSV) technology. A titanium film and a copper film are sputtered on the patterned oxide structures, which in turn are utilized as the seed layer for 20-µm copper electroplating (Figure 8(e)). The spaces between the pillars are filled by a copper electroplating process (Figure 8(f)). The electroplated copper and the PECVD-SiO₂ are planarized by a chemical mechanical polishing (CMP) process (Figure 8(g)). Then, the PECVD-SiO₂ layer is dry etched again to fabricate the via holes and the oxide hard mask (Figure 8(h), (i)). An aluminum electrode layer is deposited and dry etched, then the boundaries of the electrodes are defined (Figure 8(j), (k)). In this step, the electrodes patterns are a few micrometers (2 µm in our case) larger than that of the



oxide pillars under the electrodes, and the difference in the pattern sizes determine the length of the overhang structures. After patterning the electrode layer, the silicon substrate is etched by a DRIE process from the backside (Figure 8(l)). The copper sacrificial layer and the seed layers are removed by a wet etching process (Figure 8(m)). Finally, the fabrication process is completed by penetrating the loading slot through a DRIE process (Figure 8(n)). Figure 9 shows the fabrication results of the copper sacrificial layer method. The vertical sidewalls of oxide pillars are straight (verticality is limited by the dry etching anisotropy of the 14- μm thick SiO₂), and the overhang length is controlled to 2 μm .

Future directions

Junction ion trap

As discussed in Development history of Paul trap, the number of ion qubits trapped in an ion trap array inevitably must increase in order to adapt more complex quantum algorithms [12]. For trapping and manipulating large numbers of ions, a multi-zone ion trap composed by a number of ion trap arrays is proposed. In this multi-zone ion trap, the trapping zones are connected by "X" or "Y" junctions, and the information stored in ions can be transferred from one zone to another through the junctions. For shuttling the ions in an axial direction, the location of DC null point is moved along the axial direction by applying

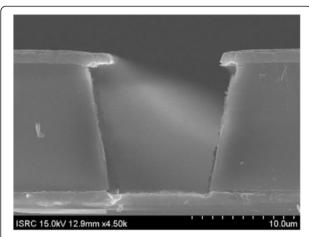


Figure 9 Scanning electron micrograph of the ion trap fabricated by the copper sacrificial layer method. The vertical sidewall of oxide pillars does not have a jagged edge and 2- μ m overhangs are shown.

time-dependent potentials to the outer DC control electrodes. Ion transports via junctions however require not only applying DC control voltages, but more complex techniques, because pseudopotential barriers created by RF voltages exist near the center of the junctions. Therefore, the geometries near the junctions should be optimized by an iterative algorithm to minimize the magnitude of the pseudopotential barriers.

Selective ion transports in junctions have also been demonstrated by a few research groups recently. Hensinger et al. and Blakestad et al. have presented ion transports in T-junction [73] and X-junction [74,75], respectively. The junction ion traps used in these experiments were built by conventional machining methods. Amini et al. [67] and Moehring et al. [76] presented ion transport experiments in Y-junctions. Wright et al. [77] has reported experimental results of ion transports in an X-junction. Figure 10 shows the Y-junction design of the National Institute of Standards and Technology (NIST) group [67]. Figure 10(a) presents an optimized design of RF electrodes at the center of junction and Figure 10(b) shows a charge coupled

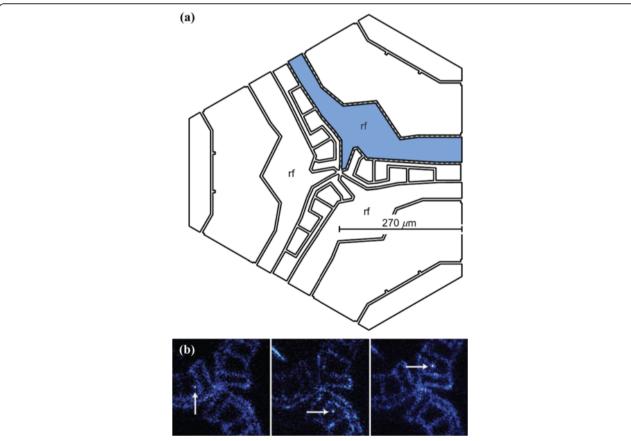


Figure 10 Y-junction surface ion trap of national institute of standards and technology (NIST) group [67]. (a) An optimized design of RF electrodes at the center of junction. **(b)** Charge coupled device (CCD) images of selective ion transports through a Y-junction geometry. White arrows point the location of the transported ion.

device (CCD) image of selective ion transports through the Y-junction geometry.

3-D ion trap fabricated by microfabrication technology

Some researchers attempted to develop 3-D ion traps using semiconductor microfabrication processes, because the confining potential of 3-D traps can be much larger and can extend the life time of trapped ions remarkably. For instance, a simulated trap depth of 3-D ion trap fabricated by microfabrication is over 10 eV [78] whereas that of surface trap is approximately $0.1 \sim 0.2$ eV [59]. The trap depth is the electric potential difference between the potential minimum point and the ion escaping point.

Using a microfabrication process, Wilpers *et al.* [79,80] have fabricated a 3-D ion trap which has a similar structure to a conventional 4-rod Paul trap. The pseudopotential shape of this microfabricated 3-D ion trap is identical to a 4-rod trap, because the distances between electrodes of the trap in horizontal and vertical directions are equal. Shaikh *et al.* [81] have developed a 3-D ion trap which has a different structure from typical 4-rod traps and higher trap depth than surface traps. Figure 11(a) shows a cross section schematic of the 3-D ion trap of National Physical Laboratory (NPL) group [79]. Figure 11(b) is a scanning electron micrograph (SEM) image of the trap. As mentioned previously, these 3-D traps can provide much larger trap depths.

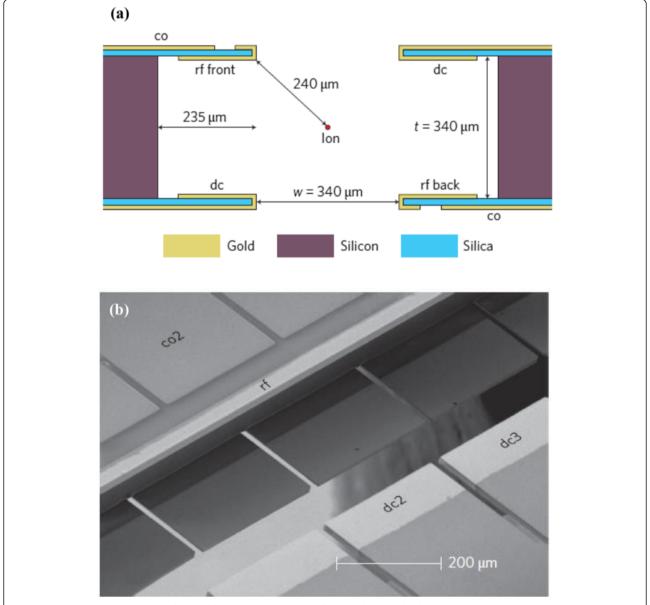


Figure 11 Schematic and picture of microfabricated 3-D ion trap of National Physical Laboratory (NPL) group [79]. (a) Cross section schematic and radial dimensions. (b) Scanning electron micrograph of the trap.

However, issues concerning a poor laser access and the geometrical complications in achieving ion shuttling need to be addressed.

Conclusion

This paper reviewed the operation principles and the development history of ion traps. Ion trap has a huge potential to be used in quantum information processing and computing. By applying MEMS-based microfabrication methods as well as conventional machining techniques, various ion traps for QIP experiments have been built and demonstrated. This paper also showed two variations of MEMS fabrication method for surface ion traps. It is expected that the ion trap technology can contribute to realize novel quantum information processing methods with exponential speed-up that we have never experienced so far. It is also expected and anticipated that MEMS fabrication technologies will be crucially instrumental in realizing complex yet inexpensive ion traps for quantum information processing and computing.

Endnote

^aPenning trap: The Penning Trap was named after F. M. Penning by Hans Georg Dehmelt because Dehmelt stated getting the inspiration of the trap from the vacuum gauge built by F. M. Penning [38].

Abbreviations

AlGaAs: Aluminum gallium arsenide; NIST: National Institute of Standards and Technology; BHF: Buffered hydrogen fluoride; NMR: Nuclear magnetic resonance; BOE: Buffered oxide etching; NPL: National Physical Laboratory; CCD: Charge coupled device; PECVD: Plasma enhanced chemical vapor deposition; CMP: Chemical mechanical polishing; PR: Photoresist; DRIE: Deep reactive ion etching; QED: Quantum electrodynamics; EMCCD: Electron multiplying charge coupled device; QIP: Quantum information processing; GaAs: Gallium arsenide; SEM: Scanning electron micrograph; LPCVD: Low pressure chemical vapor deposition; SNL: Sandia National Laboratory; MEMS: Microelectromechanical system; TSV: Through silicon via.

Competing interests

The authors declare that they have no competing interests.

Authors' contributions

DC wrote the manuscript. SH and ML surveyed the literature on quantum information processing and microfabricated ion traps. TK surveyed the literature on quantum physics and quantum computing. All authors read and approved the final manuscript.

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